

# 1,3-Dioxane, 2,4-pentyl, 2R,4R

<b>Inchi:</b>	InChI=1S/C14H28O2/c1-3-5-7-9-13-11-12-15-14(16-13)10-8-6-4-2/h13-14H,3-12H2,1-2H
<b>InchiKey:</b>	JTGGXBHDIVDFHP-KBPBESRZSA-N
<b>Formula:</b>	C14H28O2
<b>SMILES:</b>	CCCCC1CCOC(CCCCC)O1
<b>Mol. weight [g/mol]:</b>	228.37

## Physical Properties

Property code	Value	Unit	Source
gf	-88.50	kJ/mol	Joback Method
hf	-562.31	kJ/mol	Joback Method
hfus	40.88	kJ/mol	Joback Method
hvap	55.90	kJ/mol	Joback Method
log10ws	-4.47		Crippen Method
logp	4.279		Crippen Method
mcvol	209.000	ml/mol	McGowan Method
pc	1724.59	kPa	Joback Method
ripol	1847.00		NIST Webbook
ripol	1847.00		NIST Webbook
tb	588.50	K	Joback Method
tc	774.71	K	Joback Method
tf	303.82	K	Joback Method
vc	0.793	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	572.41	J/molxK	588.50	Joback Method
cpg	666.00	J/molxK	743.68	Joback Method
cpg	649.17	J/molxK	712.64	Joback Method
cpg	631.42	J/molxK	681.61	Joback Method
cpg	612.73	J/molxK	650.57	Joback Method
cpg	593.06	J/molxK	619.54	Joback Method
cpg	681.92	J/molxK	774.71	Joback Method
dvisc	0.0002092	Paxs	588.50	Joback Method

dvisc	0.0002801	Paxs	541.05	Joback Method
dvisc	0.0003967	Paxs	493.61	Joback Method
dvisc	0.0006050	Paxs	446.16	Joback Method
dvisc	0.0010203	Paxs	398.71	Joback Method
dvisc	0.0019812	Paxs	351.27	Joback Method
dvisc	0.0047334	Paxs	303.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R191776&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R191776&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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