

# 2-Methylbutyl 8-methylnon-6-enoate

<b>Inchi:</b>	InChI=1S/C15H28O2/c1-5-14(4)12-17-15(16)11-9-7-6-8-10-13(2)3/h8,10,13-14H,5-7,9,1
<b>InchiKey:</b>	PLTRJLWIJLCEOL-CSKARUKUSA-N
<b>Formula:</b>	C15H28O2
<b>SMILES:</b>	CCC(C)COC(=O)CCCCC=CC(C)C
<b>Mol. weight [g/mol]:</b>	240.38

## Physical Properties

Property code	Value	Unit	Source
gf	-83.16	kJ/mol	Joback Method
hf	-491.07	kJ/mol	Joback Method
hfus	30.55	kJ/mol	Joback Method
hvap	57.32	kJ/mol	Joback Method
log10ws	-4.33		Crippen Method
logp	4.348		Crippen Method
mvol	225.350	ml/mol	McGowan Method
pc	1545.13	kPa	Joback Method
rinpol	1545.20		NIST Webbook
rinpol	1545.20		NIST Webbook
tb	622.17	K	Joback Method
tc	800.65	K	Joback Method
tf	295.89	K	Joback Method
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	601.34	J/molxK	622.17	Joback Method
cpg	681.63	J/molxK	770.90	Joback Method
cpg	667.11	J/molxK	741.15	Joback Method
cpg	651.84	J/molxK	711.41	Joback Method
cpg	635.80	J/molxK	681.66	Joback Method
cpg	618.98	J/molxK	651.92	Joback Method
cpg	695.44	J/molxK	800.65	Joback Method
dvisc	0.0001019	Paxs	622.17	Joback Method

dvisc	0.0001417	Paxs	567.79	Joback Method
dvisc	0.0002114	Paxs	513.41	Joback Method
dvisc	0.0003466	Paxs	459.03	Joback Method
dvisc	0.0006492	Paxs	404.65	Joback Method
dvisc	0.0014772	Paxs	350.27	Joback Method
dvisc	0.0045474	Paxs	295.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U412897&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U412897&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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