

# Isophthalic acid, ethyl 3-methylbut-2-yl ester

<b>Inchi:</b>	InChI=1S/C15H20O4/c1-5-18-14(16)12-7-6-8-13(9-12)15(17)19-11(4)10(2)3/h6-11H,5H2
<b>InchiKey:</b>	FARIRUQNOWBQQI-UHFFFAOYSA-N
<b>Formula:</b>	C15H20O4
<b>SMILES:</b>	CCOC(=O)c1cccc(C(=O)OC(C)C(C)C)c1
<b>Mol. weight [g/mol]:</b>	264.32

## Physical Properties

Property code	Value	Unit	Source
gf	-294.52	kJ/mol	Joback Method
hf	-628.03	kJ/mol	Joback Method
hfus	26.79	kJ/mol	Joback Method
hvap	69.46	kJ/mol	Joback Method
log10ws	-3.92		Crippen Method
logp	3.065		Crippen Method
mvol	213.330	ml/mol	McGowan Method
pc	2003.70	kPa	Joback Method
rinpol	1908.00		NIST Webbook
rinpol	1908.00		NIST Webbook
tb	725.96	K	Joback Method
tc	936.20	K	Joback Method
tf	412.07	K	Joback Method
vc	0.803	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	594.71	J/molxK	725.96	Joback Method
cpg	610.01	J/molxK	761.00	Joback Method
cpg	624.28	J/molxK	796.04	Joback Method
cpg	637.56	J/molxK	831.08	Joback Method
cpg	649.83	J/molxK	866.12	Joback Method
cpg	661.12	J/molxK	901.16	Joback Method
cpg	671.43	J/molxK	936.20	Joback Method
dvisc	0.0012644	Paxs	412.07	Joback Method

dvisc	0.0006354	Paxs	464.39	Joback Method
dvisc	0.0003670	Paxs	516.70	Joback Method
dvisc	0.0002345	Paxs	569.02	Joback Method
dvisc	0.0001616	Paxs	621.33	Joback Method
dvisc	0.0001180	Paxs	673.64	Joback Method
dvisc	0.0000901	Paxs	725.96	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344717&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344717&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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