

# Fumaric acid, 2-methoxyphenyl 2-methylpent-3-yl ester

<b>Inchi:</b>	InChI=1S/C17H22O5/c1-5-13(12(2)3)21-16(18)10-11-17(19)22-15-9-7-6-8-14(15)20-4/h6
<b>InchiKey:</b>	RJFOZJKPZIFEEV-ZHACJKMWSA-N
<b>Formula:</b>	C17H22O5
<b>SMILES:</b>	CCC(OC(=O)C=CC(=O)Oc1ccccc1OC)C(C)C
<b>Mol. weight [g/mol]:</b>	306.35

## Physical Properties

Property code	Value	Unit	Source
gf	-302.46	kJ/mol	Joback Method
hf	-684.31	kJ/mol	Joback Method
hfus	33.36	kJ/mol	Joback Method
hvap	76.28	kJ/mol	Joback Method
log10ws	-3.84		Crippen Method
logp	3.135		Crippen Method
mcvol	243.080	ml/mol	McGowan Method
pc	1749.21	kPa	Joback Method
rinpol	2144.00		NIST Webbook
rinpol	2144.00		NIST Webbook
tb	798.30	K	Joback Method
tc	1009.29	K	Joback Method
tf	451.76	K	Joback Method
vc	0.913	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	708.20	J/mol×K	798.30	Joback Method
cpg	723.09	J/mol×K	833.47	Joback Method
cpg	736.87	J/mol×K	868.63	Joback Method
cpg	749.56	J/mol×K	903.80	Joback Method
cpg	761.18	J/mol×K	938.96	Joback Method
cpg	771.74	J/mol×K	974.13	Joback Method
cpg	781.25	J/mol×K	1009.29	Joback Method
dvisc	0.0006771	Paxs	451.76	Joback Method

dvisc	0.0003350	Paxs	509.52	Joback Method
dvisc	0.0001913	Paxs	567.27	Joback Method
dvisc	0.0001212	Paxs	625.03	Joback Method
dvisc	0.0000829	Paxs	682.79	Joback Method
dvisc	0.0000602	Paxs	740.54	Joback Method
dvisc	0.0000458	Paxs	798.30	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405929&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405929&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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