

Pentane-2,4-dione, 3-(4-methoxybenzoyl)-

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|-----------------------------|--|
| Inchi: | InChI=1S/C13H14O4/c1-8(14)12(9(2)15)13(16)10-4-6-11(17-3)7-5-10/h4-7,12H,1-3H3 |
| InchiKey: | CRQNKYNTSKUTDE-UHFFFAOYSA-N |
| Formula: | C13H14O4 |
| SMILES: | COc1ccc(C(=O)C(C(C)=O)C(C)=O)cc1 |
| Mol. weight [g/mol]: | 234.25 |
| CAS: | 137833-36-4 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -332.84 | kJ/mol | Joback Method |
| hf | -561.83 | kJ/mol | Joback Method |
| hfus | 25.54 | kJ/mol | Joback Method |
| hvap | 69.73 | kJ/mol | Joback Method |
| ie | 7.86 | eV | NIST Webbook |
| log10ws | -2.24 | | Crippen Method |
| logp | 1.672 | | Crippen Method |
| mcvol | 180.850 | ml/mol | McGowan Method |
| pc | 2576.72 | kPa | Joback Method |
| tb | 712.09 | K | Joback Method |
| tc | 933.07 | K | Joback Method |
| tf | 432.23 | K | Joback Method |
| vc | 0.685 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 477.39 | J/molxK | 712.09 | Joback Method |
| cpg | 532.90 | J/molxK | 896.24 | Joback Method |
| cpg | 523.60 | J/molxK | 859.41 | Joback Method |
| cpg | 513.41 | J/molxK | 822.58 | Joback Method |
| cpg | 502.33 | J/molxK | 785.75 | Joback Method |
| cpg | 490.33 | J/molxK | 748.92 | Joback Method |
| cpg | 541.35 | J/molxK | 933.07 | Joback Method |
| dvisc | 0.0001594 | Paxs | 712.09 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0002022 | Paxs | 665.45 | Joback Method |
| dvisc | 0.0002658 | Paxs | 618.80 | Joback Method |
| dvisc | 0.0003654 | Paxs | 572.16 | Joback Method |
| dvisc | 0.0005315 | Paxs | 525.52 | Joback Method |
| dvisc | 0.0008316 | Paxs | 478.87 | Joback Method |
| dvisc | 0.0014331 | Paxs | 432.23 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C137833364&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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