

# 1-(Prop-2-enyloxy)-2-(prop-2-enyl)-4,5-methylenecyclohexane

<b>Other names:</b>	5-Allyl-6-(allyloxy)-1,3-benzodioxole
<b>Inchi:</b>	InChI=1S/C13H14O3/c1-3-5-10-7-12-13(16-9-15-12)8-11(10)14-6-4-2/h3-4,7-8H,1-2,5-6,9-13H
<b>InchiKey:</b>	UITAOVXQEWVMGA-UHFFFAOYSA-N
<b>Formula:</b>	C13H14O3
<b>SMILES:</b>	C=CCOc1cc2c(cc1CC=C)OCO2
<b>Mol. weight [g/mol]:</b>	218.25
<b>CAS:</b>	259138-58-4

## Physical Properties

Property code	Value	Unit	Source
gf	109.00	kJ/mol	Joback Method
hf	-161.75	kJ/mol	Joback Method
hfus	33.95	kJ/mol	Joback Method
hvap	59.11	kJ/mol	Joback Method
log10ws	-3.57		Crippen Method
logp	2.709		Crippen Method
mcvol	168.420	ml/mol	McGowan Method
pc	2581.96	kPa	Joback Method
rinpol	1705.00		NIST Webbook
rinpol	1705.00		NIST Webbook
rinpol	1705.00		NIST Webbook
tb	619.55	K	Joback Method
tc	836.70	K	Joback Method
tf	394.28	K	Joback Method
vc	0.635	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	424.99	J/molxK	619.55	Joback Method
cpg	438.99	J/molxK	655.74	Joback Method
cpg	452.12	J/molxK	691.93	Joback Method
cpg	464.43	J/molxK	728.13	Joback Method
cpg	475.97	J/molxK	764.32	Joback Method

cpg	486.79	J/molxK	800.51	Joback Method
cpg	496.95	J/molxK	836.70	Joback Method
dvisc	0.0013798	Paxs	394.28	Joback Method
dvisc	0.0009714	Paxs	431.83	Joback Method
dvisc	0.0007234	Paxs	469.37	Joback Method
dvisc	0.0005627	Paxs	506.91	Joback Method
dvisc	0.0004532	Paxs	544.46	Joback Method
dvisc	0.0003753	Paxs	582.00	Joback Method
dvisc	0.0003180	Paxs	619.55	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C259138584&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C259138584&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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