

3,6-dihydroxy-megastigm-7-en-9-one

Inchi:	InChI=1S/C13H22O3/c1-9-7-11(15)8-12(3,4)13(9,16)6-5-10(2)14/h5-6,9,11,15-16H,7-8H
InchiKey:	CWLKAPCFJXJCEO-AATRIKPKSA-N
Formula:	C13H22O3
SMILES:	CC(=O)C=CC1(O)C(C)CC(O)CC1(C)C
Mol. weight [g/mol]:	226.31

Physical Properties

Property code	Value	Unit	Source
gf	-273.42	kJ/mol	Joback Method
hf	-587.69	kJ/mol	Joback Method
hfus	21.86	kJ/mol	Joback Method
hvap	81.79	kJ/mol	Joback Method
log10ws	-2.56		Crippen Method
logp	1.680		Crippen Method
mvol	192.180	ml/mol	McGowan Method
pc	2574.11	kPa	Joback Method
ripol	3118.00		NIST Webbook
ripol	3118.00		NIST Webbook
tb	745.25	K	Joback Method
tc	942.41	K	Joback Method
tf	445.22	K	Joback Method
vc	0.714	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	587.14	J/molxK	745.25	Joback Method
cpg	602.16	J/molxK	778.11	Joback Method
cpg	616.96	J/molxK	810.97	Joback Method
cpg	631.69	J/molxK	843.83	Joback Method
cpg	646.51	J/molxK	876.69	Joback Method
cpg	661.56	J/molxK	909.55	Joback Method
cpg	677.01	J/molxK	942.41	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R332922&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripl:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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