

# Carbamodithioic acid, dipropyl-, propyl ester

<b>Other names:</b>	Propyl dipropyldithiocarbamate S-Propyl-N,N-dipropyldithiocarbamate Carbamic acid, dipropyldithio-, propyl ester
<b>Inchi:</b>	InChI=1S/C10H21NS2/c1-4-7-11(8-5-2)10(12)13-9-6-3/h4-9H2,1-3H3
<b>InchiKey:</b>	IQMQTTJSYGOQRS-UHFFFAOYSA-N
<b>Formula:</b>	C10H21NS2
<b>SMILES:</b>	CCCSC(=S)N(CCC)CCC
<b>Mol. weight [g/mol]:</b>	219.41
<b>CAS:</b>	19047-79-1

## Physical Properties

Property code	Value	Unit	Source
gf	294.28	kJ/mol	Joback Method
hf	6.17	kJ/mol	Joback Method
hfus	33.41	kJ/mol	Joback Method
hvap	53.44	kJ/mol	Joback Method
log10ws	-3.81		Crippen Method
logp	3.537		Crippen Method
mcvol	190.140	ml/mol	McGowan Method
pc	2331.52	kPa	Joback Method
rinpol	1652.00		NIST Webbook
rinpol	1750.00		NIST Webbook
rinpol	1652.00		NIST Webbook
tb	579.46	K	Joback Method
tc	781.46	K	Joback Method
tf	303.60	K	Joback Method
vc	0.704	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	452.25	J/mol×K	579.46	Joback Method
cpg	467.73	J/mol×K	613.13	Joback Method
cpg	482.32	J/mol×K	646.79	Joback Method

cpg	496.08	J/mol×K	680.46	Joback Method
cpg	509.05	J/mol×K	714.13	Joback Method
cpg	521.29	J/mol×K	747.79	Joback Method
cpg	532.84	J/mol×K	781.46	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19047791&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19047791&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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