

Sodium phosphate, tribasic-

Other names:	trisodium phosphate
Inchi:	InChI=1S/3Na.H3O4P/c;;;1-5(2,3)4/h;;;(H3,1,2,3,4)/q3*+1;/p-3
InchiKey:	RYFMWSXOAZQYPI-UHFFFAOYSA-K
Formula:	Na3O4P
SMILES:	O=P([O-])([O-])[O-].[Na+].[Na].[Na]
Mol. weight [g/mol]:	163.94

Sources

Sodium salts solubility in ternary glycerol+water+alcohol mixtures Phase Behavior for the Aqueous Two-Phase Systems Containing the Ethanol of CO₂ in aqueous TSP: 1-Butyl-3-methylimidazolium VoluMetric IsoThermic Compressibility Tetrafluoroborate and KBrO₄ Propanol and electrical conductivity of solutions of 1-butyl-3-methylimidazolium propanol volumetric and compressibility over the temperature range of 298.15 to 315 K: Bromide and Hexafluorophosphate Anhydrous solutions and 298.15 to 303.15 aqueous solutions of orthophosphate Comparative study of density, sound velocity and refractive index for (water + alkyl methyl) phosphates aqueous systems at T = (298.15, 303.15, and 308.15) K: Solubilities of imidazolium-based ionic liquids in aqueous salt solutions at 298.15 K: Liquid phase equilibria in nicotine (aqueous) solutions: Phase Diagrams for Aqueous Two-Phase Systems Containing the Influence of Sodium Ions on the Miscelization and Interfacial Behavior of Cetan-1, Sulfate and Trisodium Phosphate in Aqueous Solutions: Bromide and Hexafluorophosphate in Aqueous Solution at 298.15 K: Diffusion Coefficients and Correlation

<https://www.chemeo.com/cid/93-309-6/Sodium-phosphate-tribasic.pdf>

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