

Sodium phosphate, tribasic-

Other names:	trisodium phosphate
Inchi:	InChI=1S/3Na.H3O4P/c;;1-5(2,3)4/h;;(H3,1,2,3,4)/q3*+1;/p-3
InchiKey:	RYFMWSXOAZQYPI-UHFFFAOYSA-K
Formula:	Na3O4P
SMILES:	O=P([O-])[O-].[Na+].[Na]
Mol. weight [g/mol]:	163.94

Sources

Phase Behavior for the Aqueous Two-Phase Systems Containing the Solubility of CO₂ in aqueous TSP: 1-Butyl-3-methylimidazolium Phase Diagrams for Aqueous Two-Phase Systems Containing the Solubilities of imidazolium-based ionic transfers in aqueous salt solutions by Tetra-n-butylammonium and Trisodium Phosphate/Sodium Sulfite/Sodium Dihydrogen Phosphate at -38.15 K and electrical conductivity of Solutions of sodium phosphate monopropanol volumetric and osmotic pressure over the Comparative Study of Sodium 5K: Biphase data and sodium sulfite influence on Solutions at 298.15 K 353.15 Micellization and Interfacial Behavior of Cationic Surfactant equilibria in Dispersed (aqueous) solutions Bromide in Aqueous Solution Thermodynamic properties of binary aqueous solutions of orthophosphate Comparative study of density, sound velocity and fraggance index for (water + potassium metal) solubilities in various systems at 298.15, 308.15, and 308.53 in purification process of crude glycerol from the biodiesel industry:

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- <https://www.doi.org/10.1021/je900533h>
- <https://www.doi.org/10.1016/j.fluid.2012.05.014>
- <https://www.doi.org/10.1021/je200379r>
- <https://www.doi.org/10.1016/j.jct.2011.03.002>
- <http://webbook.nist.gov/cgi/cbook.cgi?ID=B6002381&Units=SI>
- <https://www.doi.org/10.1016/j.fluid.2008.01.004>
- <https://www.doi.org/10.1016/j.jct.2013.07.027>
- <https://www.doi.org/10.1021/acs.jced.5b00203>
- <https://www.doi.org/10.1021/je501058a>
- <https://www.doi.org/10.1016/j.fluid.2011.08.022>
- <https://www.doi.org/10.1016/j.fluid.2014.02.019>
- <https://www.doi.org/10.1016/j.jct.2013.01.012>
- <https://www.doi.org/10.1016/j.fluid.2019.05.023>