

# 2,3,5-Triiodobenzyl alcohol, n-butyl ether

<b>Inchi:</b>	InChI=1S/C11H13I3O/c1-2-3-4-15-7-8-5-9(12)6-10(13)11(8)14/h5-6H,2-4,7H2,1H3
<b>InchiKey:</b>	YZCUYJDABIDFNK-UHFFFAOYSA-N
<b>Formula:</b>	C11H13I3O
<b>SMILES:</b>	CCCCOCc1cc(I)cc(I)c1I
<b>Mol. weight [g/mol]:</b>	541.93

## Physical Properties

Property code	Value	Unit	Source
gf	194.62	kJ/mol	Joback Method
hf	30.14	kJ/mol	Joback Method
hfus	31.53	kJ/mol	Joback Method
hvap	74.87	kJ/mol	Joback Method
log10ws	-6.56		Crippen Method
logp	4.817		Crippen Method
mcvol	225.420	ml/mol	McGowan Method
pc	2331.52	kPa	Joback Method
rinpol	2508.00		NIST Webbook
rinpol	2508.00		NIST Webbook
tb	794.54	K	Joback Method
tc	1077.66	K	Joback Method
tf	474.12	K	Joback Method
vc	0.826	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	462.37	J/molxK	794.54	Joback Method
cpg	473.56	J/molxK	841.73	Joback Method
cpg	483.87	J/molxK	888.91	Joback Method
cpg	493.40	J/molxK	936.10	Joback Method
cpg	502.22	J/molxK	983.29	Joback Method
cpg	510.43	J/molxK	1030.47	Joback Method
cpg	518.11	J/molxK	1077.66	Joback Method
dvisc	0.0007897	Paxs	474.12	Joback Method

dvisc	0.0004716	Paxs	527.52	Joback Method
dvisc	0.0003096	Paxs	580.93	Joback Method
dvisc	0.0002182	Paxs	634.33	Joback Method
dvisc	0.0001624	Paxs	687.73	Joback Method
dvisc	0.0001261	Paxs	741.14	Joback Method
dvisc	0.0001013	Paxs	794.54	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375219&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375219&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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