

# Lavandulyle 2-methylbutyrate

<b>Other names:</b>	Lavandulyl 2-methylbutyrate
<b>Inchi:</b>	InChI=1S/C15H26O2/c1-7-13(6)15(16)17-10-14(12(4)5)9-8-11(2)3/h13-14H,2,4,7-10H2,1
<b>InchiKey:</b>	IBKFLDWCCDDVSDC-UHFFFAOYSA-N
<b>Formula:</b>	C15H26O2
<b>SMILES:</b>	<chem>C=C(C)CCC(COC(=O)C(C)CC)C(=C)C</chem>
<b>Mol. weight [g/mol]:</b>	238.37

## Physical Properties

Property code	Value	Unit	Source
gf	-4.80	kJ/mol	Joback Method
hf	-377.01	kJ/mol	Joback Method
hfus	25.17	kJ/mol	Joback Method
hvap	56.18	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	4.124		Crippen Method
mcvol	221.050	ml/mol	McGowan Method
pc	1597.44	kPa	Joback Method
rinpol	1490.00		NIST Webbook
rinpol	1512.00		NIST Webbook
rinpol	1495.00		NIST Webbook
rinpol	1490.00		NIST Webbook
ripol	1759.00		NIST Webbook
tb	611.13	K	Joback Method
tc	793.92	K	Joback Method
tf	269.53	K	Joback Method
vc	0.852	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	577.03	J/molxK	611.13	Joback Method
cpg	594.52	J/molxK	641.60	Joback Method
cpg	611.17	J/molxK	672.06	Joback Method
cpg	627.02	J/molxK	702.53	Joback Method

cpg	642.09	J/mol×K	732.99	Joback Method
cpg	656.39	J/mol×K	763.46	Joback Method
cpg	669.96	J/mol×K	793.92	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R232783&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R232783&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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