

# 2-(2-(2-(2-Pentoxy-ethoxy)-ethoxy)-ethoxy)-ethoxy

<b>Other names:</b>	Pentaethylene glycol, pentyl ether
<b>Inchi:</b>	InChI=1S/C15H32O6/c1-2-3-4-6-17-8-10-19-12-14-21-15-13-20-11-9-18-7-5-16/h16H,2-
<b>InchiKey:</b>	CHFJFGVAPWOVGG-UHFFFAOYSA-N
<b>Formula:</b>	C15H32O6
<b>SMILES:</b>	CCCCOCCOCCOCCOCCOCCO
<b>Mol. weight [g/mol]:</b>	308.41

## Physical Properties

Property code	Value	Unit	Source
gf	-586.40	kJ/mol	Joback Method
hf	-1166.26	kJ/mol	Joback Method
hfus	44.63	kJ/mol	Joback Method
hvap	77.71	kJ/mol	Joback Method
log10ws	-0.80		Crippen Method
logp	1.252		Crippen Method
mcvol	257.430	ml/mol	McGowan Method
pc	1422.92	kPa	Joback Method
rinpol	2234.90		NIST Webbook
tb	746.88	K	Joback Method
tc	916.73	K	Joback Method
tf	430.78	K	Joback Method
vc	0.985	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	788.93	J/molxK	746.88	Joback Method
cpg	861.74	J/molxK	888.43	Joback Method
cpg	848.81	J/molxK	860.12	Joback Method
cpg	835.06	J/molxK	831.81	Joback Method
cpg	820.48	J/molxK	803.50	Joback Method
cpg	805.10	J/molxK	775.19	Joback Method
cpg	873.83	J/molxK	916.73	Joback Method
dvisc	0.0000098	Paxs	746.88	Joback Method

dvisc	0.0000147	Paxs	694.20	Joback Method
dvisc	0.0000238	Paxs	641.51	Joback Method
dvisc	0.0000419	Paxs	588.83	Joback Method
dvisc	0.0000825	Paxs	536.15	Joback Method
dvisc	0.0001879	Paxs	483.46	Joback Method
dvisc	0.0005238	Paxs	430.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R188344&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R188344&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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