

# Succinic acid, 4-chloro-3-methylphenyl but-3-en-1-yl ester

<b>Inchi:</b>	InChI=1S/C15H17ClO4/c1-3-4-9-19-14(17)7-8-15(18)20-12-5-6-13(16)11(2)10-12/h3,5-6
<b>InchiKey:</b>	NCHKAOKOADHIRM-UHFFFAOYSA-N
<b>Formula:</b>	C15H17ClO4
<b>SMILES:</b>	C=CCCOC(=O)CCC(=O)Oc1ccc(Cl)c(C)c1
<b>Mol. weight [g/mol]:</b>	296.75

## Physical Properties

Property code	Value	Unit	Source
gf	-223.36	kJ/mol	Joback Method
hf	-519.25	kJ/mol	Joback Method
hfus	36.36	kJ/mol	Joback Method
hvap	74.61	kJ/mol	Joback Method
log10ws	-4.17		Crippen Method
logp	3.453		Crippen Method
mvol	221.270	ml/mol	McGowan Method
pc	1957.87	kPa	Joback Method
rinpol	2162.00		NIST Webbook
rinpol	2162.00		NIST Webbook
tb	765.93	K	Joback Method
tc	976.90	K	Joback Method
tf	482.75	K	Joback Method
vc	0.846	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	593.68	J/molxK	765.93	Joback Method
cpg	606.71	J/molxK	801.09	Joback Method
cpg	618.82	J/molxK	836.25	Joback Method
cpg	630.02	J/molxK	871.41	Joback Method
cpg	640.31	J/molxK	906.57	Joback Method
cpg	649.72	J/molxK	941.73	Joback Method
cpg	658.25	J/molxK	976.90	Joback Method
dvisc	0.0006825	Paxs	482.75	Joback Method

dvisc	0.0004306	Paxs	529.95	Joback Method
dvisc	0.0002929	Paxs	577.14	Joback Method
dvisc	0.0002112	Paxs	624.34	Joback Method
dvisc	0.0001595	Paxs	671.54	Joback Method
dvisc	0.0001249	Paxs	718.73	Joback Method
dvisc	0.0001009	Paxs	765.93	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391199&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391199&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/93-537-3/Succinic-acid-4-chloro-3-methylphenyl-but-3-en-1-yl-ester.pdf>

Generated by Cheméo on 2024-05-04 02:25:45.791094678 +0000 UTC m=+17078794.711671993.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.