

# 1,3-Propanedione, 2-bromo-1,3-diphenyl-

<b>Other names:</b>	Bromodibenzoylmethane Dibenzoylbromomethane Methane, dibenzoylbromo- 2-Bromo-1,3-diphenyl-1,3-propanedione
<b>Inchi:</b>	InChI=1S/C15H11BrO2/c16-13(14(17)11-7-3-1-4-8-11)15(18)12-9-5-2-6-10-12/h1-10,13
<b>InchiKey:</b>	BYAJHZYXPBREEK-UHFFFAOYSA-N
<b>Formula:</b>	C15H11BrO2
<b>SMILES:</b>	O=C(c1ccccc1)C(Br)C(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	303.15
<b>CAS:</b>	728-84-7

## Physical Properties

Property code	Value	Unit	Source
gf	54.28	kJ/mol	Joback Method
hf	-83.98	kJ/mol	Joback Method
hfus	27.65	kJ/mol	Joback Method
hvap	73.08	kJ/mol	Joback Method
log10ws	-4.57		Crippen Method
logp	3.516		Crippen Method
mcvol	195.330	ml/mol	McGowan Method
pc	3110.57	kPa	Joback Method
tb	769.42	K	Joback Method
tc	1031.47	K	Joback Method
tf	456.31	K	Joback Method
vc	0.728	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	490.00	J/molxK	769.42	Joback Method
cpg	502.54	J/molxK	813.10	Joback Method
cpg	513.85	J/molxK	856.77	Joback Method
cpg	524.06	J/molxK	900.45	Joback Method
cpg	533.27	J/molxK	944.12	Joback Method

cpg	541.60	J/molxK	987.80	Joback Method
cpg	549.18	J/molxK	1031.47	Joback Method
dvisc	0.0014639	Paxs	456.31	Joback Method
dvisc	0.0008149	Paxs	508.50	Joback Method
dvisc	0.0005059	Paxs	560.68	Joback Method
dvisc	0.0003406	Paxs	612.87	Joback Method
dvisc	0.0002440	Paxs	665.05	Joback Method
dvisc	0.0001835	Paxs	717.23	Joback Method
dvisc	0.0001434	Paxs	769.42	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C728847&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C728847&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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