

# Succinic acid, 4-chloro-3-methylphenyl 2-ethoxyethyl ester

Inchi:	InChI=1S/C15H19ClO5/c1-3-19-8-9-20-14(17)6-7-15(18)21-12-4-5-13(16)11(2)10-12/h4-
InchiKey:	XIOZSGLPUVHMMA-UHFFFAOYSA-N
Formula:	C15H19ClO5
SMILES:	CCOCCOC(=O)CCC(=O)Oc1ccc(Cl)c(C)c1
Mol. weight [g/mol]:	314.76

## Physical Properties

Property code	Value	Unit	Source
gf	-416.20	kJ/mol	Joback Method
hf	-776.90	kJ/mol	Joback Method
hfus	38.83	kJ/mol	Joback Method
hvap	77.69	kJ/mol	Joback Method
log10ws	-3.41		Crippen Method
logp	2.914		Crippen Method
mvol	231.440	ml/mol	McGowan Method
pc	1859.51	kPa	Joback Method
rinpol	2263.00		NIST Webbook
rinpol	2263.00		NIST Webbook
tb	791.67	K	Joback Method
tc	998.75	K	Joback Method
tf	506.74	K	Joback Method
vc	0.882	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	645.71	J/molxK	791.67	Joback Method
cpg	658.94	J/molxK	826.18	Joback Method
cpg	671.19	J/molxK	860.70	Joback Method
cpg	682.43	J/molxK	895.21	Joback Method
cpg	692.67	J/molxK	929.73	Joback Method
cpg	701.90	J/molxK	964.24	Joback Method
cpg	710.12	J/molxK	998.75	Joback Method
dvisc	0.0004861	Paxs	506.74	Joback Method

dvisc	0.0003088	Paxs	554.23	Joback Method
dvisc	0.0002107	Paxs	601.72	Joback Method
dvisc	0.0001520	Paxs	649.20	Joback Method
dvisc	0.0001147	Paxs	696.69	Joback Method
dvisc	0.0000897	Paxs	744.18	Joback Method
dvisc	0.0000723	Paxs	791.67	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390670&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390670&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/93-634-5/Succinic-acid-4-chloro-3-methylphenyl-2-ethoxyethyl-ester.pdf>

Generated by Cheméo on 2024-04-25 14:07:06.438719124 +0000 UTC m=+16343275.359296452.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.