

# Aniline, p-(2,3-epoxypropoxy)-N,N-bis(2,3-epoxypropyl)-

<b>Other names:</b>	Oxiranemethanamine, N-[4-(oxiranylmethoxy)phenyl]-N-(oxiranylmethyl)- p-(Diglycidylamino)phenyl glycidyl ether p-Aminophenol triglycidyl ether ERL 0500 N,N,O-Tris(2,3-epoxypropyl)-p-aminophenol 4-(Diglycidylamino)phenyl glycidyl ether 2-Oxiranemethanamine, N-[4-(2-oxiranylmethoxy)phenyl]-N-(2-oxiranylmethyl)- p-(2,3-epoxypropoxy)-N,N-bis(2,3-epoxypropyl)aniline N,N,N-Glycidyl p-aminophenol
<b>Inchi:</b>	InChI=1S/C15H19NO4/c1-3-12(17-9-15-10-20-15)4-2-11(1)16(5-13-7-18-13)6-14-8-19-1
<b>InchiKey:</b>	AHIPJALLQVEEQF-UHFFFAOYSA-N
<b>Formula:</b>	C15H19NO4
<b>SMILES:</b>	<chem>c1cc(N(CC2CO2)CC2CO2)ccc1OCC1CO1</chem>
<b>Mol. weight [g/mol]:</b>	277.32
<b>CAS:</b>	5026-74-4

## Physical Properties

Property code	Value	Unit	Source
gf	107.87	kJ/mol	Joback Method
hf	-370.16	kJ/mol	Joback Method
hfs	-110.50 ± 1.40	kJ/mol	NIST Webbook
hfus	50.81	kJ/mol	Joback Method
hvap	69.64	kJ/mol	Joback Method
log10ws	-1.29		Crippen Method
logp	1.068		Crippen Method
mcvol	199.330	ml/mol	McGowan Method
pc	2505.01	kPa	Joback Method
tb	710.19	K	Joback Method
tc	932.35	K	Joback Method
tf	485.98	K	Joback Method
vc	0.738	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	617.16	J/mol×K	710.19	Joback Method
cpg	634.23	J/mol×K	747.22	Joback Method
cpg	650.12	J/mol×K	784.24	Joback Method
cpg	664.96	J/mol×K	821.27	Joback Method
cpg	678.88	J/mol×K	858.30	Joback Method
cpg	692.00	J/mol×K	895.32	Joback Method
cpg	704.46	J/mol×K	932.35	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5026744&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5026744&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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