

# 11-Dodecenoic acid

<b>Inchi:</b>	InChI=1S/C12H22O2/c1-2-3-4-5-6-7-8-9-10-11-12(13)14/h2H,1,3-11H2,(H,13,14)
<b>InchiKey:</b>	GZZPOFFXKUVNSW-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O2
<b>SMILES:</b>	C=CCCCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	198.30
<b>CAS:</b>	65423-25-8

## Physical Properties

Property code	Value	Unit	Source
gf	-127.74	kJ/mol	Joback Method
hf	-430.39	kJ/mol	Joback Method
hfus	31.24	kJ/mol	Joback Method
hvap	65.06	kJ/mol	Joback Method
log10ws	-3.80		Crippen Method
logp	3.768		Crippen Method
mvol	183.080	ml/mol	McGowan Method
pc	2117.78	kPa	Joback Method
ripol	2565.00		NIST Webbook
ripol	2565.00		NIST Webbook
tb	616.69	K	Joback Method
tc	784.80	K	Joback Method
tf	333.99	K	Joback Method
vc	0.714	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	481.13	J/mol×K	616.69	Joback Method
cpg	494.34	J/mol×K	644.71	Joback Method
cpg	506.96	J/mol×K	672.73	Joback Method
cpg	519.02	J/mol×K	700.74	Joback Method
cpg	530.52	J/mol×K	728.76	Joback Method
cpg	541.50	J/mol×K	756.78	Joback Method
cpg	551.98	J/mol×K	784.80	Joback Method

dvisc	0.0075368	Paxs	333.99	Joback Method
dvisc	0.0021882	Paxs	381.11	Joback Method
dvisc	0.0008340	Paxs	428.22	Joback Method
dvisc	0.0003849	Paxs	475.34	Joback Method
dvisc	0.0002042	Paxs	522.46	Joback Method
dvisc	0.0001203	Paxs	569.57	Joback Method
dvisc	0.0000769	Paxs	616.69	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39339e+01
Coeff. B	-4.61171e+03
Coeff. C	-9.93740e+01
Temperature range (K), min.	437.32
Temperature range (K), max.	634.23

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C65423258&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C65423258&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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