

# 7-ethoxy-3,7-dimethyl-1,5-octadien-3-ol

<b>Other names:</b>	3,7-Dimethyl-1,5-octadien-7-ethoxy-3-ol
<b>Inchi:</b>	InChI=1S/C12H22O2/c1-6-12(5,13)10-8-9-11(3,4)14-7-2/h6,8-9,13H,1,7,10H2,2-5H3/b9-
<b>InchiKey:</b>	IPSITHRBRXCRLC-CMDGGOBGSA-N
<b>Formula:</b>	C12H22O2
<b>SMILES:</b>	<chem>C=CC(C)(O)CC=CC(C)(C)OCC</chem>
<b>Mol. weight [g/mol]:</b>	198.30

## Physical Properties

Property code	Value	Unit	Source
gf	-17.92	kJ/mol	Joback Method
hf	-350.31	kJ/mol	Joback Method
hfus	16.21	kJ/mol	Joback Method
hvap	58.09	kJ/mol	Joback Method
log10ws	-3.13		Crippen Method
logp	2.685		Crippen Method
mcvol	183.080	ml/mol	McGowan Method
pc	2137.41	kPa	Joback Method
ripol	1725.00		NIST Webbook
tb	582.94	K	Joback Method
tc	764.96	K	Joback Method
tf	306.05	K	Joback Method
vc	0.683	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	472.21	J/molxK	582.94	Joback Method
cpg	487.03	J/molxK	613.28	Joback Method
cpg	501.01	J/molxK	643.61	Joback Method
cpg	514.20	J/molxK	673.95	Joback Method
cpg	526.66	J/molxK	704.29	Joback Method
cpg	538.42	J/molxK	734.63	Joback Method
cpg	549.54	J/molxK	764.96	Joback Method
dvisc	0.0139126	Paxs	306.05	Joback Method

dvisc	0.0028857	Paxs	352.20	Joback Method
dvisc	0.0008617	Paxs	398.35	Joback Method
dvisc	0.0003307	Paxs	444.50	Joback Method
dvisc	0.0001520	Paxs	490.64	Joback Method
dvisc	0.0000798	Paxs	536.79	Joback Method
dvisc	0.0000464	Paxs	582.94	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R324144&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R324144&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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