

# Undecanoic acid, 10-oxo-, methyl ester

<b>Other names:</b>	Methyl 10-oxoundecanoate
<b>Inchi:</b>	InChI=1S/C12H22O3/c1-11(13)9-7-5-3-4-6-8-10-12(14)15-2/h3-10H2,1-2H3
<b>InchiKey:</b>	TXRMRSKJQBFMRQ-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O3
<b>SMILES:</b>	COC(=O)CCCCCCCCC(C)=O
<b>Mol. weight [g/mol]:</b>	214.30
<b>CAS:</b>	18993-09-4

## Physical Properties

Property code	Value	Unit	Source
gf	-312.68	kJ/mol	Joback Method
hf	-648.39	kJ/mol	Joback Method
hfus	31.22	kJ/mol	Joback Method
hvap	58.21	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	2.869		Crippen Method
mcvol	188.950	ml/mol	McGowan Method
pc	1956.14	kPa	Joback Method
rinpol	1572.00		NIST Webbook
ripol	2200.00		NIST Webbook
tb	604.12	K	Joback Method
tc	781.95	K	Joback Method
tf	347.09	K	Joback Method
vc	0.738	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	488.75	J/molxK	604.12	Joback Method
cpg	503.49	J/molxK	633.76	Joback Method
cpg	517.57	J/molxK	663.40	Joback Method
cpg	531.02	J/molxK	693.04	Joback Method
cpg	543.84	J/molxK	722.67	Joback Method
cpg	556.04	J/molxK	752.31	Joback Method

cpg	567.63	J/molxK	781.95	Joback Method
dvisc	0.0024044	Paxs	347.09	Joback Method
dvisc	0.0012547	Paxs	389.93	Joback Method
dvisc	0.0007447	Paxs	432.77	Joback Method
dvisc	0.0004856	Paxs	475.61	Joback Method
dvisc	0.0003398	Paxs	518.44	Joback Method
dvisc	0.0002511	Paxs	561.28	Joback Method
dvisc	0.0001937	Paxs	604.12	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18993094&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18993094&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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