

# Cyclohexanecarboxylic acid, 4-methoxy-, butyl ester

Inchi:	InChI=1S/C12H22O3/c1-3-4-9-15-12(13)10-5-7-11(14-2)8-6-10/h10-11H,3-9H2,1-2H3
InchiKey:	RWMKKRUZLLRKQE-UHFFFAOYSA-N
Formula:	C12H22O3
SMILES:	CCCCOC(=O)C1CCC(OC)CC1
Mol. weight [g/mol]:	214.30

## Physical Properties

Property code	Value	Unit	Source
gf	-272.02	kJ/mol	Joback Method
hf	-634.05	kJ/mol	Joback Method
hfus	23.72	kJ/mol	Joback Method
hvap	53.99	kJ/mol	Joback Method
log10ws	-2.56		Crippen Method
logp	2.535		Crippen Method
mcvol	182.390	ml/mol	McGowan Method
pc	2125.60	kPa	Joback Method
rinsol	1542.00		NIST Webbook
tb	587.55	K	Joback Method
tc	785.10	K	Joback Method
tf	322.53	K	Joback Method
vc	0.681	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	484.16	J/molxK	587.55	Joback Method
cpg	503.22	J/molxK	620.48	Joback Method
cpg	521.36	J/molxK	653.40	Joback Method
cpg	538.58	J/molxK	686.33	Joback Method
cpg	554.87	J/molxK	719.25	Joback Method
cpg	570.24	J/molxK	752.18	Joback Method
cpg	584.69	J/molxK	785.10	Joback Method
dvisc	0.0022926	Paxs	322.53	Joback Method
dvisc	0.0011673	Paxs	366.70	Joback Method

dvisc	0.0006872	Paxs	410.87	Joback Method
dvisc	0.0004484	Paxs	455.04	Joback Method
dvisc	0.0003155	Paxs	499.21	Joback Method
dvisc	0.0002351	Paxs	543.38	Joback Method
dvisc	0.0001831	Paxs	587.55	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406191&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406191&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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