

# Cyanoacetic acid, 1-methylcyclohexyl, ethyl ester

<b>Other names:</b>	Acetic acid, 2-cyano-2-(1-methylcyclohexyl), ethyl ester
<b>Inchi:</b>	InChI=1S/C12H19NO2/c1-3-15-11(14)10(9-13)12(2)7-5-4-6-8-12/h10H,3-8H2,1-2H3
<b>InchiKey:</b>	ROHCCJIABRKISQ-UHFFFAOYSA-N
<b>Formula:</b>	C12H19NO2
<b>SMILES:</b>	CCOC(=O)C(C#N)C1(C)CCCCC1
<b>Mol. weight [g/mol]:</b>	209.28

## Physical Properties

Property code	Value	Unit	Source
gf	-34.06	kJ/mol	Joback Method
hf	-306.65	kJ/mol	Joback Method
hfus	13.14	kJ/mol	Joback Method
hvap	60.83	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	2.660		Crippen Method
mcvol	177.900	ml/mol	McGowan Method
pc	2287.14	kPa	Joback Method
rinpol	1489.00		NIST Webbook
rinpol	1498.00		NIST Webbook
rinpol	1508.00		NIST Webbook
rinpol	1489.00		NIST Webbook
ripol	2005.00		NIST Webbook
ripol	2050.00		NIST Webbook
ripol	2005.00		NIST Webbook
ripol	2095.00		NIST Webbook
tb	671.68	K	Joback Method
tc	896.77	K	Joback Method
tf	378.43	K	Joback Method
vc	0.682	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	491.98	J/molxK	671.68	Joback Method

cpg	508.34	J/mol×K	709.20	Joback Method
cpg	523.86	J/mol×K	746.71	Joback Method
cpg	538.64	J/mol×K	784.23	Joback Method
cpg	552.82	J/mol×K	821.74	Joback Method
cpg	566.49	J/mol×K	859.26	Joback Method
cpg	579.79	J/mol×K	896.77	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R97705&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R97705&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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