

# 2,4,6-Tribromobiphenyl

<b>Inchi:</b>	InChI=1S/C12H7Br3/c13-9-6-10(14)12(11(15)7-9)8-4-2-1-3-5-8/h1-7H
<b>InchiKey:</b>	YCLZBYQDXVJFII-UHFFFAOYSA-N
<b>Formula:</b>	C12H7Br3
<b>SMILES:</b>	BrC1cc(Br)c(-c2ccccc2)c(Br)c1
<b>Mol. weight [g/mol]:</b>	390.90
<b>CAS:</b>	59080-33-0

## Physical Properties

Property code	Value	Unit	Source
gf	289.05	kJ/mol	Joback Method
hf	226.63	kJ/mol	Joback Method
hfus	29.61	kJ/mol	Joback Method
hvap	68.15	kJ/mol	Joback Method
log10ws	-7.30		Aqueous Solubility Prediction Method
logp	5.641		Crippen Method
mcvol	184.920	ml/mol	McGowan Method
pc	4183.90	kPa	Joback Method
tb	740.74	K	Joback Method
tc	1031.10	K	Joback Method
tf	494.80	K	Joback Method
vc	0.677	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	373.57	J/molxK	740.74	Joback Method
cpg	416.95	J/molxK	982.71	Joback Method
cpg	409.64	J/molxK	934.31	Joback Method
cpg	401.79	J/molxK	885.92	Joback Method
cpg	393.25	J/molxK	837.53	Joback Method
cpg	383.89	J/molxK	789.13	Joback Method
cpg	423.87	J/molxK	1031.10	Joback Method
dvisc	0.0001631	Paxs	740.74	Joback Method

dvisc	0.0001942	Paxs	699.75	Joback Method
dvisc	0.0002364	Paxs	658.76	Joback Method
dvisc	0.0002952	Paxs	617.77	Joback Method
dvisc	0.0003806	Paxs	576.78	Joback Method
dvisc	0.0005101	Paxs	535.79	Joback Method
dvisc	0.0007176	Paxs	494.80	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C59080330&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C59080330&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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