

4-Bromo-2-fluorobiphenyl

Other names:	1,1'-Biphenyl, 4-bromo-2-fluoro-4-bromo-2-fluoro-1,1'-biphenyl
Inchi:	InChI=1S/C12H8BrF/c13-10-6-7-11(12(14)8-10)9-4-2-1-3-5-9/h1-8H
InchiKey:	HTRNHWBOBYFTQF-UHFFFAOYSA-N
Formula:	C12H8BrF
SMILES:	Fc1cc(Br)ccc1-c1ccccc1
Mol. weight [g/mol]:	251.09
CAS:	41604-19-7

Physical Properties

Property code	Value	Unit	Source
gf	75.23	kJ/mol	Joback Method
hf	-10.67	kJ/mol	Joback Method
hfus	22.51	kJ/mol	Joback Method
hvap	53.80	kJ/mol	Joback Method
log10ws	-5.57		Crippen Method
logp	4.255		Crippen Method
mcvol	151.690	ml/mol	McGowan Method
pc	3403.91	kPa	Joback Method
tb	602.71	K	Joback Method
tc	855.66	K	Joback Method
tf	363.27	K	Joback Method
vc	0.572	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	323.16	J/mol×K	602.71	Joback Method
cpg	336.38	J/mol×K	644.87	Joback Method
cpg	348.49	J/mol×K	687.03	Joback Method
cpg	359.58	J/mol×K	729.19	Joback Method
cpg	369.71	J/mol×K	771.34	Joback Method
cpg	378.97	J/mol×K	813.50	Joback Method
cpg	387.43	J/mol×K	855.66	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C41604197&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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