

# 10H-Phenothiazine, 2-chloro-

<b>Other names:</b>	2-chloro-10H-phenothiazine 2-chlorophenothiazine Chlorpromazine M (ring) Phenothiazine, 2-chloro-
<b>Inchi:</b>	InChI=1S/C12H8CINS/c13-8-5-6-12-10(7-8)14-9-3-1-2-4-11(9)15-12/h1-7,14H
<b>InchiKey:</b>	KFZGLJSYQXZIGP-UHFFFAOYSA-N
<b>Formula:</b>	C12H8CINS
<b>SMILES:</b>	Clc1ccc2c(c1)Nc1cccc1S2
<b>Mol. weight [g/mol]:</b>	233.72
<b>CAS:</b>	92-39-7

## Physical Properties

Property code	Value	Unit	Source
gf	442.29	kJ/mol	Joback Method
hf	314.27	kJ/mol	Joback Method
hfus	30.36	kJ/mol	Joback Method
hvap	65.85	kJ/mol	Joback Method
log10ws	-4.62		Crippen Method
logp	4.548		Crippen Method
mcvol	160.130	ml/mol	McGowan Method
pc	3749.97	kPa	Joback Method
rinpol	2099.00		NIST Webbook
rinpol	2099.00		NIST Webbook
rinpol	2099.00		NIST Webbook
rinpol	2099.00		NIST Webbook
tb	683.21	K	Joback Method
tc	963.50	K	Joback Method
tf	472.95	K	Solubility determination and thermodynamic functions of 2-chlorophenothiazine in nine organic solvents from T = 283.15 K to T = 318.15 K and mixing properties of solutions
vc	0.590	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	363.10	J/molxK	683.21	Joback Method
cpg	375.23	J/molxK	729.93	Joback Method
cpg	386.30	J/molxK	776.64	Joback Method
cpg	396.47	J/molxK	823.36	Joback Method
cpg	405.88	J/molxK	870.07	Joback Method
cpg	414.69	J/molxK	916.79	Joback Method
cpg	423.05	J/molxK	963.50	Joback Method

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C92397&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C92397&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Solubility determination and thermodynamic functions of 2-thiazolethione in nine organic solvents from T = 283.15 K to T = 318.15 K and mixing properties of solutions:	<a href="https://www.doi.org/10.1016/j.jct.2016.11.029">https://www.doi.org/10.1016/j.jct.2016.11.029</a>
Joback Method	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/94-027-8/10H-Phenothiazine-2-chloro.pdf>

Generated by Cheméo on 2024-04-23 19:30:00.449513436 +0000 UTC m=+16189849.370090751.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.