

# 2-Propenoic acid, 3-(3,4,5-trimethoxyphenyl)-, methyl ester, cis

Inchi:	InChI=1S/C12H14O4/c1-14-10-6-4-9(8-11(10)15-2)5-7-12(13)16-3/h4-8H,1-3H3/b7-5-
InchiKey:	JXRYDOZRPYFBKO-ALCCZGGFSA-N
Formula:	C12H14O4
SMILES:	COC(=O)C=Cc1ccc(OC)c(OC)c1
Mol. weight [g/mol]:	222.24

## Physical Properties

Property code	Value	Unit	Source
gf	-220.39	kJ/mol	Joback Method
hf	-469.44	kJ/mol	Joback Method
hfus	25.46	kJ/mol	Joback Method
hvap	59.84	kJ/mol	Joback Method
log10ws	-2.23		Crippen Method
logp	1.890		Crippen Method
mcvol	171.060	ml/mol	McGowan Method
pc	2487.55	kPa	Joback Method
rinsol	1851.00		NIST Webbook
tb	635.89	K	Joback Method
tc	847.87	K	Joback Method
tf	388.00	K	Joback Method
vc	0.639	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	422.99	J/molxK	635.89	Joback Method
cpg	436.66	J/molxK	671.22	Joback Method
cpg	449.59	J/molxK	706.55	Joback Method
cpg	461.77	J/molxK	741.88	Joback Method
cpg	473.21	J/molxK	777.21	Joback Method
cpg	483.89	J/molxK	812.54	Joback Method
cpg	493.81	J/molxK	847.87	Joback Method
dvisc	0.0006990	Paxs	388.00	Joback Method
dvisc	0.0004299	Paxs	429.31	Joback Method

dvisc	0.0002880	Paxs	470.63	Joback Method
dvisc	0.0002058	Paxs	511.94	Joback Method
dvisc	0.0001546	Paxs	553.26	Joback Method
dvisc	0.0001209	Paxs	594.58	Joback Method
dvisc	0.0000976	Paxs	635.89	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R298436&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R298436&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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