

# 2,3,5-Triiodobenzyl alcohol, neopentyl ether

<b>Inchi:</b>	InChI=1S/C12H15I3O/c1-12(2,3)7-16-6-8-4-9(13)5-10(14)11(8)15/h4-5H,6-7H2,1-3H3
<b>InchiKey:</b>	BIVLTHOMYUYGJA-UHFFFAOYSA-N
<b>Formula:</b>	C12H15I3O
<b>SMILES:</b>	CC(C)(C)COc1cc(I)cc(I)c1I
<b>Mol. weight [g/mol]:</b>	555.96

## Physical Properties

Property code	Value	Unit	Source
gf	205.88	kJ/mol	Joback Method
hf	0.75	kJ/mol	Joback Method
hfus	26.70	kJ/mol	Joback Method
hvap	75.80	kJ/mol	Joback Method
log10ws	-6.73		Crippen Method
logp	5.063		Crippen Method
mcvol	239.510	ml/mol	McGowan Method
pc	2167.36	kPa	Joback Method
rinpol	2466.00		NIST Webbook
rinpol	2466.00		NIST Webbook
tb	814.19	K	Joback Method
tc	1106.78	K	Joback Method
tf	487.81	K	Joback Method
vc	0.871	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	516.84	J/mol×K	814.19	Joback Method
cpg	528.90	J/mol×K	862.96	Joback Method
cpg	540.01	J/mol×K	911.72	Joback Method
cpg	550.35	J/mol×K	960.49	Joback Method
cpg	560.05	J/mol×K	1009.25	Joback Method
cpg	569.28	J/mol×K	1058.02	Joback Method
cpg	578.18	J/mol×K	1106.78	Joback Method
dvisc	0.0006701	Paxs	487.81	Joback Method

dvisc	0.0003807	Paxs	542.21	Joback Method
dvisc	0.0002398	Paxs	596.60	Joback Method
dvisc	0.0001632	Paxs	651.00	Joback Method
dvisc	0.0001178	Paxs	705.40	Joback Method
dvisc	0.0000891	Paxs	759.79	Joback Method
dvisc	0.0000700	Paxs	814.19	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375231&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375231&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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