

# Benzamide, 3-methyl-N-butyl-

Inchi:	InChI=1S/C12H17NO/c1-3-4-8-13-12(14)11-7-5-6-10(2)9-11/h5-7,9H,3-4,8H2,1-2H3,(H,14)
InchiKey:	NXDWJMPJVGIQBB-UHFFFAOYSA-N
Formula:	C12H17NO
SMILES:	CCCCN=C(O)c1ccccc(C)c1
Mol. weight [g/mol]:	191.27

## Physical Properties

Property code	Value	Unit	Source
hf	-145.75	kJ/mol	Joback Method
hvap	65.32	kJ/mol	Joback Method
log10ws	-3.09		Crippen Method
logp	3.100		Crippen Method
mcvol	167.730	ml/mol	McGowan Method
pc	2351.92	kPa	Joback Method
rinpol	1766.00		NIST Webbook
tb	674.36	K	Joback Method
tc	880.11	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407411&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407411&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

Latest version available from:

<https://www.chemeo.com/cid/94-087-2/Benzamide-3-methyl-N-butyl.pdf>

Generated by Cheméo on 2024-04-19 17:07:03.511089557 +0000 UTC m=+15835672.431666869.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.