

# **((CH3)2N)2C=N-(4-CH3O-C6H4)**

**Inchi:** InChI=1S/C12H19N3O/c1-14(2)12(15(3)4)13-10-6-8-11(16-5)9-7-10/h6-9H,1-5H3  
**InchiKey:** IHEVQCXBPONXGO-UHFFFAOYSA-N  
**Formula:** C12H19N3O  
**SMILES:** COc1ccc(N=C(N(C)C)N(C)C)cc1  
**Mol. weight [g/mol]:** 221.30  
**CAS:** 20815-35-4

## Physical Properties

Property code	Value	Unit	Source
affp	1047.70	kJ/mol	NIST Webbook
basg	1015.20	kJ/mol	NIST Webbook
hf	9.32	kJ/mol	Joback Method
hvap	55.13	kJ/mol	Joback Method
log10ws	-1.55		Crippen Method
logp	1.806		Crippen Method
mcvol	187.690	ml/mol	McGowan Method
pc	2113.89	kPa	Joback Method
tb	629.48	K	Joback Method
tc	841.84	K	Joback Method

## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C20815354&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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