

N-(beta-dimethylaminoethyl)-o-phenylenediamine

Inchi:	InChI=1S/C10H17N3/c1-13(2)8-7-12-10-6-4-3-5-9(10)11/h3-6,12H,7-8,11H2,1-2H3
InchiKey:	YMPYPZBOBAISUFL-UHFFFAOYSA-N
Formula:	C10H17N3
SMILES:	CN(C)CCNc1ccccc1N
Mol. weight [g/mol]:	179.26
CAS:	78156-03-3

Physical Properties

Property code	Value	Unit	Source
gf	402.72	kJ/mol	Joback Method
hf	130.12	kJ/mol	Joback Method
hfus	28.63	kJ/mol	Joback Method
hvap	59.91	kJ/mol	Joback Method
log10ws	-0.94		Crippen Method
logp	1.242		Crippen Method
mcvol	157.940	ml/mol	McGowan Method
pc	3107.10	kPa	Joback Method
tb	595.00	K	Joback Method
tc	808.51	K	Joback Method
tf	409.79	K	Joback Method
vc	0.570	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	401.19	J/mol×K	595.00	Joback Method
cpg	416.19	J/mol×K	630.58	Joback Method
cpg	430.25	J/mol×K	666.17	Joback Method
cpg	443.41	J/mol×K	701.75	Joback Method
cpg	455.72	J/mol×K	737.34	Joback Method
cpg	467.22	J/mol×K	772.92	Joback Method
cpg	477.96	J/mol×K	808.51	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C78156033&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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