

# Diethylmalonic acid, 4-bromo-2-methoxyphenyl heptyl ester

<b>Inchi:</b>	InChI=1S/C21H31BrO5/c1-5-8-9-10-11-14-26-19(23)21(6-2,7-3)20(24)27-17-13-12-16(2
<b>InchiKey:</b>	QWDVFMQFYJNLFO-UHFFFAOYSA-N
<b>Formula:</b>	C21H31BrO5
<b>SMILES:</b>	CCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(Br)cc1OC
<b>Mol. weight [g/mol]:</b>	443.37

## Physical Properties

Property code	Value	Unit	Source
gf	-336.59	kJ/mol	Joback Method
hf	-867.42	kJ/mol	Joback Method
hfus	48.04	kJ/mol	Joback Method
hvap	91.80	kJ/mol	Joback Method
log10ws	-6.71		Crippen Method
logp	5.683		Crippen Method
mvol	321.240	ml/mol	McGowan Method
pc	1303.29	kPa	Joback Method
rinpol	2633.00		NIST Webbook
rinpol	2633.00		NIST Webbook
tb	954.45	K	Joback Method
tc	1173.86	K	Joback Method
tf	606.66	K	Joback Method
vc	1.220	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1004.99	J/molxK	954.45	Joback Method
cpg	1018.97	J/molxK	991.02	Joback Method
cpg	1031.67	J/molxK	1027.59	Joback Method
cpg	1043.14	J/molxK	1064.16	Joback Method
cpg	1053.40	J/molxK	1100.72	Joback Method
cpg	1062.51	J/molxK	1137.29	Joback Method
cpg	1070.49	J/molxK	1173.86	Joback Method
dvisc	0.0001848	Paxs	606.66	Joback Method

dvisc	0.0001101	Paxs	664.62	Joback Method
dvisc	0.0000713	Paxs	722.59	Joback Method
dvisc	0.0000492	Paxs	780.56	Joback Method
dvisc	0.0000358	Paxs	838.52	Joback Method
dvisc	0.0000271	Paxs	896.49	Joback Method
dvisc	0.0000212	Paxs	954.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370951&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370951&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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