

# Benzeneacetic acid, 4-hydroxy-3-methoxy-, methyl ester

<b>Other names:</b>	Acetic acid, (4-hydroxy-3-methoxyphenyl)-, methyl ester Homovanillic acid methyl ester Methyl 3-methoxy-4-hydroxyphenylacetate Methyl homovanillate methyl 4-hydroxy-3-methoxyphenylacetate
<b>Inchi:</b>	InChI=1S/C10H12O4/c1-13-9-5-7(3-4-8(9)11)6-10(12)14-2/h3-5,11H,6H2,1-2H3
<b>InchiKey:</b>	JJJSFAGPWHEUBT-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O4
<b>SMILES:</b>	<chem>COC(=O)Cc1ccc(O)c(OC)c1</chem>
<b>Mol. weight [g/mol]:</b>	196.20
<b>CAS:</b>	15964-80-4

## Physical Properties

Property code	Value	Unit	Source
gf	-357.44	kJ/mol	Joback Method
hf	-579.00	kJ/mol	Joback Method
hfus	25.07	kJ/mol	Joback Method
hvap	65.37	kJ/mol	Joback Method
log10ws	-1.22		Crippen Method
logp	1.116		Crippen Method
mcvol	147.180	ml/mol	McGowan Method
pc	3526.27	kPa	Joback Method
tb	639.19	K	Joback Method
tc	860.68	K	Joback Method
tf	447.51	K	Joback Method
vc	0.495	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	372.77	J/mol×K	639.19	Joback Method
cpg	384.49	J/mol×K	676.10	Joback Method
cpg	395.53	J/mol×K	713.02	Joback Method
cpg	405.91	J/mol×K	749.93	Joback Method

cpg	415.69	J/molxK	786.85	Joback Method
cpg	424.90	J/molxK	823.76	Joback Method
cpg	433.57	J/molxK	860.68	Joback Method
dvisc	0.0003718	Paxs	447.51	Joback Method
dvisc	0.0001978	Paxs	479.46	Joback Method
dvisc	0.0001139	Paxs	511.40	Joback Method
dvisc	0.0000700	Paxs	543.35	Joback Method
dvisc	0.0000454	Paxs	575.30	Joback Method
dvisc	0.0000308	Paxs	607.24	Joback Method
dvisc	0.0000217	Paxs	639.19	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15964804&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15964804&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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