

N-benzyl-n-methylamino-acetonitrile

Inchi:	InChI=1S/C10H12N2/c1-12(8-7-11)9-10-5-3-2-4-6-10/h2-6H,8-9H2,1H3
InchiKey:	KUSXDIGALTYDLG-UHFFFAOYSA-N
Formula:	C10H12N2
SMILES:	CN(CC#N)Cc1ccccc1
Mol. weight [g/mol]:	160.22
CAS:	14321-25-6

Physical Properties

Property code	Value	Unit	Source
gf	389.69	kJ/mol	Joback Method
hf	219.21	kJ/mol	Joback Method
hfus	20.22	kJ/mol	Joback Method
hvap	52.65	kJ/mol	Joback Method
log10ws	-2.04		Crippen Method
logp	1.642		Crippen Method
mcvol	139.360	ml/mol	McGowan Method
pc	2847.48	kPa	Joback Method
tb	569.40	K	Joback Method
tc	789.25	K	Joback Method
tf	326.34	K	Joback Method
vc	0.531	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	322.93	J/molxK	569.40	Joback Method
cpg	336.12	J/molxK	606.04	Joback Method
cpg	348.41	J/molxK	642.68	Joback Method
cpg	359.85	J/molxK	679.32	Joback Method
cpg	370.50	J/molxK	715.96	Joback Method
cpg	380.39	J/molxK	752.60	Joback Method
cpg	389.58	J/molxK	789.25	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C14321256&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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