

# Succinic acid, dec-2-yl 4-chloro-2-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C21H31ClO5/c1-4-5-6-7-8-9-10-16(2)26-20(23)13-14-21(24)27-18-12-11-17(2)
<b>InchiKey:</b>	QGDBHFJCLWWOKX-UHFFFAOYSA-N
<b>Formula:</b>	C21H31ClO5
<b>SMILES:</b>	CCCCCCCCC(C)OC(=O)CCC(=O)Oc1ccc(Cl)cc1OC
<b>Mol. weight [g/mol]:</b>	398.92

## Physical Properties

Property code	Value	Unit	Source
gf	-368.12	kJ/mol	Joback Method
hf	-906.02	kJ/mol	Joback Method
hfus	50.84	kJ/mol	Joback Method
hvap	90.66	kJ/mol	Joback Method
log10ws	-6.59		Crippen Method
logp	5.716		Crippen Method
mvol	315.980	ml/mol	McGowan Method
pc	1199.79	kPa	Joback Method
rinpol	2753.00		NIST Webbook
rinpol	2753.00		NIST Webbook
tb	928.51	K	Joback Method
tc	1140.15	K	Joback Method
tf	559.36	K	Joback Method
vc	1.212	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	992.54	J/molxK	928.51	Joback Method
cpg	1051.36	J/molxK	1104.88	Joback Method
cpg	1042.30	J/molxK	1069.60	Joback Method
cpg	1031.90	J/molxK	1034.33	Joback Method
cpg	1020.15	J/molxK	999.06	Joback Method
cpg	1007.03	J/molxK	963.78	Joback Method
cpg	1059.09	J/molxK	1140.15	Joback Method
dvisc	0.0000282	Paxs	928.51	Joback Method

dvisc	0.0000362	Paxs	866.98	Joback Method
dvisc	0.0000484	Paxs	805.46	Joback Method
dvisc	0.0000678	Paxs	743.93	Joback Method
dvisc	0.0001010	Paxs	682.41	Joback Method
dvisc	0.0001628	Paxs	620.88	Joback Method
dvisc	0.0002914	Paxs	559.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390943&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390943&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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