

# 1-Hydroxy-2-(prop-2-enyl)-4,5-methylenedioxybenzene

<b>Inchi:</b>	InChI=1S/C10H10O3/c1-2-3-7-4-9-10(5-8(7)11)13-6-12-9/h2,4-5,11H,1,3,6H2
<b>InchiKey:</b>	HSDKTWGTSNDEBZ-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O3
<b>SMILES:</b>	C=CCc1cc2c(cc1O)OCO2
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	19202-23-4

## Physical Properties

Property code	Value	Unit	Source
gf	-44.09	kJ/mol	Joback Method
hf	-258.88	kJ/mol	Joback Method
hfus	32.44	kJ/mol	Joback Method
hvap	63.04	kJ/mol	Joback Method
log10ws	-2.31		Crippen Method
logp	1.849		Crippen Method
mcvol	130.450	ml/mol	McGowan Method
pc	4211.09	kPa	Joback Method
rinpol	1547.00		NIST Webbook
tb	607.45	K	Joback Method
tc	844.46	K	Joback Method
tf	439.20	K	Joback Method
vc	0.434	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	328.75	J/molxK	607.45	Joback Method
cpg	377.38	J/molxK	804.96	Joback Method
cpg	368.87	J/molxK	765.46	Joback Method
cpg	359.88	J/molxK	725.95	Joback Method
cpg	350.28	J/molxK	686.45	Joback Method
cpg	339.95	J/molxK	646.95	Joback Method
cpg	385.54	J/molxK	844.46	Joback Method
dvisc	0.0000755	Paxs	607.45	Joback Method

dvisc	0.0001024	Paxs	579.41	Joback Method
dvisc	0.0001431	Paxs	551.37	Joback Method
dvisc	0.0002074	Paxs	523.33	Joback Method
dvisc	0.0003134	Paxs	495.28	Joback Method
dvisc	0.0004976	Paxs	467.24	Joback Method
dvisc	0.0008382	Paxs	439.20	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19202234&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19202234&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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