

# p-(Dimethylamino)cinnamic acid

**Other names:**

4-(N,N-Dimethylamino)cinnamic acid  
4-Dimethylamino cinnamic acid  
Dimethylaminocinnamic acid  
2-Propenoic acid, 3-[4-(dimethylamino)phenyl]-  
4-(N-Dimethylamino)cinnamic acid  
3-(p-(Dimethylamino)phenyl)acrylic acid

**Inchi:**

InChI=1S/C11H13NO2/c1-12(2)10-6-3-9(4-7-10)5-8-11(13)14/h3-8H,1-2H3,(H,13,14)/b8

**InchiKey:**

CQNPVMCASGWEHM-VMPITWQZSA-N

**Formula:**

C<sub>11</sub>H<sub>13</sub>NO<sub>2</sub>

**SMILES:**

CN(C)c1ccc(C=CC(=O)O)cc1

**Mol. weight [g/mol]:**

191.23

**CAS:**

1552-96-1

## Physical Properties

Property code	Value	Unit	Source
gf	69.78	kJ/mol	Joback Method
hf	-125.37	kJ/mol	Joback Method
hfus	26.81	kJ/mol	Joback Method
hvap	68.44	kJ/mol	Joback Method
log10ws	-1.72		Crippen Method
logp	1.850		Crippen Method
mcvol	155.210	ml/mol	McGowan Method
pc	3254.14	kPa	Joback Method
tb	645.39	K	Joback Method
tc	849.07	K	Joback Method
tf	390.81	K	Joback Method
vc	0.567	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	389.64	J/mol×K	645.39	Joback Method
cpg	401.25	J/mol×K	679.34	Joback Method
cpg	412.11	J/mol×K	713.28	Joback Method

cpg	422.25	J/mol×K	747.23	Joback Method
cpg	431.75	J/mol×K	781.18	Joback Method
cpg	440.63	J/mol×K	815.12	Joback Method
cpg	448.95	J/mol×K	849.07	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1552961&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1552961&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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