

# Phenol, 2-ethoxy-5-(1-propenyl)-

<b>Other names:</b>	1-Ethoxy-2-hydroxy-4-propenylbenzene Propenylguaethol trans-2-Ethoxy-5-(1-propenyl)phenol Hydroxy methyl anethol Isosafroegenol Phenol, 2-ethoxy-5-propenyl- Vanitrope 2-Ethoxy-5-propenylphenol 6-Ethoxy-m-anol 2-ethoxy-5-prop-1-enylphenol
<b>Inchi:</b>	InChI=1S/C11H14O2/c1-3-5-9-6-7-11(13-4-2)10(12)8-9/h3,5-8,12H,4H2,1-2H3/b5-3+
<b>InchiKey:</b>	RADIRXJQODWKGQ-HWKANZROSA-N
<b>Formula:</b>	C11H14O2
<b>SMILES:</b>	CC=Cc1ccc(OCC)c(O)c1
<b>Mol. weight [g/mol]:</b>	178.23
<b>CAS:</b>	94-86-0

## Physical Properties

Property code	Value	Unit	Source
gf	-34.88	kJ/mol	Joback Method
hf	-237.62	kJ/mol	Joback Method
hfus	25.07	kJ/mol	Joback Method
hvap	58.40	kJ/mol	Joback Method
log10ws	-2.80		Crippen Method
logp	2.824		Crippen Method
mcvol	149.530	ml/mol	McGowan Method
pc	3177.55	kPa	Joback Method
tb	589.94	K	Joback Method
tc	812.56	K	Joback Method
tf	381.54	K	Joback Method
vc	0.507	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	361.75	J/mol×K	589.94	Joback Method
cpg	375.14	J/mol×K	627.04	Joback Method
cpg	387.68	J/mol×K	664.15	Joback Method
cpg	399.45	J/mol×K	701.25	Joback Method
cpg	410.54	J/mol×K	738.35	Joback Method
cpg	421.01	J/mol×K	775.46	Joback Method
cpg	430.94	J/mol×K	812.56	Joback Method
dvisc	0.0010233	Paxs	381.54	Joback Method
dvisc	0.0004324	Paxs	416.27	Joback Method
dvisc	0.0002086	Paxs	451.01	Joback Method
dvisc	0.0001117	Paxs	485.74	Joback Method
dvisc	0.0000650	Paxs	520.47	Joback Method
dvisc	0.0000405	Paxs	555.21	Joback Method
dvisc	0.0000267	Paxs	589.94	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C94860&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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