

# Succinic acid, 8-chlorooctyl non-3-en-1-yl ester

Inchi:	InChI=1S/C21H37ClO4/c1-2-3-4-5-7-10-13-18-25-20(23)15-16-21(24)26-19-14-11-8-6-9
InchiKey:	SMJMRHXMSVIFLZ-JXMROGBWSA-N
Formula:	C21H37ClO4
SMILES:	CCCCC=CCCOC(=O)CCC(=O)OCCCCCCCCCl
Mol. weight [g/mol]:	388.97

## Physical Properties

Property code	Value	Unit	Source
gf	-273.61	kJ/mol	Joback Method
hf	-864.89	kJ/mol	Joback Method
hfus	60.12	kJ/mol	Joback Method
hvap	84.99	kJ/mol	Joback Method
log10ws	-6.34		Crippen Method
logp	5.959		Crippen Method
mvol	329.570	ml/mol	McGowan Method
pc	1022.69	kPa	Joback Method
rinpol	2775.00		NIST Webbook
rinpol	2775.00		NIST Webbook
tb	874.05	K	Joback Method
tc	1070.61	K	Joback Method
tf	495.59	K	Joback Method
vc	1.288	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1035.08	J/molxK	874.05	Joback Method
cpg	1052.31	J/molxK	906.81	Joback Method
cpg	1068.44	J/molxK	939.57	Joback Method
cpg	1083.52	J/molxK	972.33	Joback Method
cpg	1097.59	J/molxK	1005.09	Joback Method
cpg	1110.66	J/molxK	1037.85	Joback Method
cpg	1122.79	J/molxK	1070.61	Joback Method
dvisc	0.0005469	Paxs	495.59	Joback Method

dvisc	0.0002655	Paxs	558.67	Joback Method
dvisc	0.0001493	Paxs	621.74	Joback Method
dvisc	0.0000933	Paxs	684.82	Joback Method
dvisc	0.0000631	Paxs	747.90	Joback Method
dvisc	0.0000454	Paxs	810.97	Joback Method
dvisc	0.0000342	Paxs	874.05	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391102&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391102&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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