

# Fumaric acid, naphth-1-yl 8-chlorooctyl ester

<b>Inchi:</b>	InChI=1S/C22H25ClO4/c23-16-7-3-1-2-4-8-17-26-21(24)14-15-22(25)27-20-13-9-11-18-
<b>InchiKey:</b>	CMWUKCGKDORCPR-CCEZHUSRSA-N
<b>Formula:</b>	C22H25ClO4
<b>SMILES:</b>	O=C(C=CC(=O)Oc1cccc2cccc12)OCCCCCCCCCl
<b>Mol. weight [g/mol]:</b>	388.88

## Physical Properties

Property code	Value	Unit	Source
gf	-55.76	kJ/mol	Joback Method
hf	-469.40	kJ/mol	Joback Method
hfus	53.38	kJ/mol	Joback Method
hvap	91.80	kJ/mol	Joback Method
log10ws	-6.65		Crippen Method
logp	5.424		Crippen Method
mvol	300.440	ml/mol	McGowan Method
pc	1411.18	kPa	Joback Method
rinpol	3090.00		NIST Webbook
rinpol	3090.00		NIST Webbook
tb	947.57	K	Joback Method
tc	1170.47	K	Joback Method
tf	578.50	K	Joback Method
vc	1.159	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	915.83	J/molxK	947.57	Joback Method
cpg	929.41	J/molxK	984.72	Joback Method
cpg	942.03	J/molxK	1021.87	Joback Method
cpg	953.77	J/molxK	1059.02	Joback Method
cpg	964.71	J/molxK	1096.17	Joback Method
cpg	974.93	J/molxK	1133.32	Joback Method
cpg	984.51	J/molxK	1170.47	Joback Method
dvisc	0.0004628	Paxs	578.50	Joback Method

dvisc	0.0002835	Paxs	640.01	Joback Method
dvisc	0.0001892	Paxs	701.52	Joback Method
dvisc	0.0001348	Paxs	763.03	Joback Method
dvisc	0.0001010	Paxs	824.55	Joback Method
dvisc	0.0000788	Paxs	886.06	Joback Method
dvisc	0.0000635	Paxs	947.57	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405817&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405817&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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