

# 1,2-Cyclohexanedicarboxylic acid, hexadecyl 4-octyl ester

Inchi:	InChI=1S/C32H60O4/c1-4-7-9-10-11-12-13-14-15-16-17-18-19-22-27-35-31(33)29-25-20
InchiKey:	BNTGPKZCEXLQFE-UHFFFAOYSA-N
Formula:	C32H60O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C1CCCCC1C(=O)OC(CCC)CCCC
Mol. weight [g/mol]:	508.82

## Physical Properties

Property code	Value	Unit	Source
gf	-234.98	kJ/mol	Joback Method
hf	-1164.71	kJ/mol	Joback Method
hfus	73.59	kJ/mol	Joback Method
hvap	104.87	kJ/mol	Joback Method
log10ws	-10.47		Crippen Method
logp	9.719		Crippen Method
mvol	465.760	ml/mol	McGowan Method
pc	619.41	kPa	Joback Method
rinpol	3652.00		NIST Webbook
rinpol	3652.00		NIST Webbook
tb	1098.58	K	Joback Method
tc	1372.59	K	Joback Method
tf	582.86	K	Joback Method
vc	1.802	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1734.74	J/molxK	1098.58	Joback Method
cpg	1813.23	J/molxK	1326.92	Joback Method
cpg	1803.04	J/molxK	1281.25	Joback Method
cpg	1790.23	J/molxK	1235.58	Joback Method
cpg	1774.66	J/molxK	1189.92	Joback Method
cpg	1756.20	J/molxK	1144.25	Joback Method
cpg	1820.92	J/molxK	1372.59	Joback Method
dvisc	0.0000105	Paxs	1098.58	Joback Method

dvisc	0.0000143	Paxs	1012.63	Joback Method
dvisc	0.0000205	Paxs	926.67	Joback Method
dvisc	0.0000319	Paxs	840.72	Joback Method
dvisc	0.0000546	Paxs	754.77	Joback Method
dvisc	0.0001074	Paxs	668.81	Joback Method
dvisc	0.0002579	Paxs	582.86	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339530&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339530&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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