

# Urs-12-en-28-al, 3-(acetyloxy)-, (3«beta»)-

<b>Other names:</b>	3«beta»-Acetoxyurs-12-en-28-al
<b>Inchi:</b>	InChI=1S/C32H50O3/c1-20-11-16-32(19-33)18-17-30(7)23(27(32)21(20)2)9-10-25-29(6)
<b>InchiKey:</b>	RVSZZPCRCJYAEU-UHFFFAOYSA-N
<b>Formula:</b>	C32H50O3
<b>SMILES:</b>	CC(=O)OC1CCC2(C)C(CCC3(C)C2CC=C2C4C(C)C(C)CCC4(C=O)CCC23C)C1(C)C
<b>Mol. weight [g/mol]:</b>	482.74
<b>CAS:</b>	86996-88-5

## Physical Properties

Property code	Value	Unit	Source
gf	50.79	kJ/mol	Joback Method
hf	-712.84	kJ/mol	Joback Method
hfus	35.46	kJ/mol	Joback Method
hvap	96.82	kJ/mol	Joback Method
log10ws	-8.38		Crippen Method
logp	7.775		Crippen Method
mcvol	412.150	ml/mol	McGowan Method
pc	919.95	kPa	Joback Method
rinpol	3727.00		NIST Webbook
rinpol	3727.00		NIST Webbook
tb	1097.42	K	Joback Method
tc	1351.73	K	Joback Method
tf	736.96	K	Joback Method
vc	1.567	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1756.33	J/molxK	1097.42	Joback Method
cpg	1823.77	J/molxK	1139.80	Joback Method
cpg	1897.89	J/molxK	1182.19	Joback Method
cpg	1979.56	J/molxK	1224.57	Joback Method
cpg	2069.62	J/molxK	1266.96	Joback Method
cpg	2168.96	J/molxK	1309.34	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C86996885&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C86996885&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/94-708-2/Urs-12-en-28-al-3-acetyloxy-3-beta.pdf>

Generated by Cheméo on 2024-04-19 18:15:19.12182761 +0000 UTC m=+15839768.042404932.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.