

# Succinic acid, 2-(2-chlorophenoxy)ethyl pentadecyl ester

Inchi:	InChI=1S/C27H43ClO5/c1-2-3-4-5-6-7-8-9-10-11-12-13-16-21-32-26(29)19-20-27(30)33
InchiKey:	IZJQRBOXIOJPGF-UHFFFAOYSA-N
Formula:	C27H43ClO5
SMILES:	CCCCCCCCCCCCCOC(=O)CCC(=O)OCCOc1ccccc1Cl
Mol. weight [g/mol]:	483.08

## Physical Properties

Property code	Value	Unit	Source
gf	-305.53	kJ/mol	Joback Method
hf	-1013.11	kJ/mol	Joback Method
hfus	70.30	kJ/mol	Joback Method
hvap	103.74	kJ/mol	Joback Method
log10ws	-8.37		Crippen Method
logp	7.677		Crippen Method
mvol	400.520	ml/mol	McGowan Method
pc	837.24	kPa	Joback Method
rinpol	3448.00		NIST Webbook
rinpol	3448.00		NIST Webbook
tb	1061.25	K	Joback Method
tc	1307.13	K	Joback Method
tf	629.46	K	Joback Method
vc	1.554	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1359.83	J/molxK	1061.25	Joback Method
cpg	1375.49	J/molxK	1102.23	Joback Method
cpg	1389.10	J/molxK	1143.21	Joback Method
cpg	1400.74	J/molxK	1184.19	Joback Method
cpg	1410.46	J/molxK	1225.17	Joback Method
cpg	1418.32	J/molxK	1266.15	Joback Method
cpg	1424.37	J/molxK	1307.13	Joback Method
dvisc	0.0001450	Paxs	629.46	Joback Method

dvisc	0.0000772	Paxs	701.43	Joback Method
dvisc	0.0000462	Paxs	773.39	Joback Method
dvisc	0.0000302	Paxs	845.36	Joback Method
dvisc	0.0000211	Paxs	917.32	Joback Method
dvisc	0.0000155	Paxs	989.28	Joback Method
dvisc	0.0000119	Paxs	1061.25	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381544&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381544&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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