

# Succinic acid, naphth-2-ylmethyl 3-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C21H17ClO4/c22-18-6-3-7-19(13-18)26-21(24)11-10-20(23)25-14-15-8-9-16-4
<b>InchiKey:</b>	FCOQAFJOMJBLAO-UHFFFAOYSA-N
<b>Formula:</b>	C21H17ClO4
<b>SMILES:</b>	O=C(CCC(=O)Oc1cccc(Cl)c1)OCc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	368.81

## Physical Properties

Property code	Value	Unit	Source
gf	-41.62	kJ/mol	Joback Method
hf	-340.92	kJ/mol	Joback Method
hfus	44.24	kJ/mol	Joback Method
hvap	92.55	kJ/mol	Joback Method
log10ws	-6.50		Crippen Method
logp	4.922		Crippen Method
mvol	266.890	ml/mol	McGowan Method
pc	1887.08	kPa	Joback Method
rinpol	3152.00		NIST Webbook
rinpol	3152.00		NIST Webbook
tb	952.19	K	Joback Method
tc	1196.11	K	Joback Method
tf	611.25	K	Joback Method
vc	1.014	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	778.89	J/molxK	952.19	Joback Method
cpg	790.25	J/molxK	992.84	Joback Method
cpg	800.48	J/molxK	1033.50	Joback Method
cpg	809.67	J/molxK	1074.15	Joback Method
cpg	817.88	J/molxK	1114.81	Joback Method
cpg	825.21	J/molxK	1155.46	Joback Method
cpg	831.74	J/molxK	1196.11	Joback Method
dvisc	0.0004743	Paxs	611.25	Joback Method

dvisc	0.0003184	Paxs	668.07	Joback Method
dvisc	0.0002276	Paxs	724.90	Joback Method
dvisc	0.0001708	Paxs	781.72	Joback Method
dvisc	0.0001333	Paxs	838.54	Joback Method
dvisc	0.0001073	Paxs	895.37	Joback Method
dvisc	0.0000887	Paxs	952.19	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389861&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389861&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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