

# 1,3-Propanediol, docosyl ethyl ether

Inchi:	InChI=1S/C27H56O2/c1-3-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23-25-29
InchiKey:	ITIOIGVBVLGYEE-UHFFFAOYSA-N
Formula:	C27H56O2
SMILES:	CCCCCCCCCCCCCCCCCCCCOCCCOCC
Mol. weight [g/mol]:	412.73

## Physical Properties

Property code	Value	Unit	Source
gf	-33.54	kJ/mol	Joback Method
hf	-865.05	kJ/mol	Joback Method
hfus	68.06	kJ/mol	Joback Method
hvap	80.52	kJ/mol	Joback Method
log10ws	-9.30		Crippen Method
logp	9.252		Crippen Method
mcvol	403.030	ml/mol	McGowan Method
pc	676.76	kPa	Joback Method
rinpol	2771.00		NIST Webbook
rinpol	2771.00		NIST Webbook
tb	862.00	K	Joback Method
tc	1058.81	K	Joback Method
tf	438.51	K	Joback Method
vc	1.583	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1354.41	J/molxK	862.00	Joback Method
cpg	1464.42	J/molxK	1026.01	Joback Method
cpg	1445.19	J/molxK	993.21	Joback Method
cpg	1424.62	J/molxK	960.40	Joback Method
cpg	1402.66	J/molxK	927.60	Joback Method
cpg	1379.27	J/molxK	894.80	Joback Method
cpg	1482.34	J/molxK	1058.81	Joback Method
dvisc	0.0000210	Paxs	862.00	Joback Method

dvisc	0.0000289	Paxs	791.42	Joback Method
dvisc	0.0000424	Paxs	720.84	Joback Method
dvisc	0.0000676	Paxs	650.25	Joback Method
dvisc	0.0001208	Paxs	579.67	Joback Method
dvisc	0.0002533	Paxs	509.09	Joback Method
dvisc	0.0006744	Paxs	438.51	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406356&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406356&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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