

# Benzoic acid, 4-tert-butyl-, pentadecyl ester

**Inchi:** InChI=1S/C26H44O2/c1-5-6-7-8-9-10-11-12-13-14-15-16-17-22-28-25(27)23-18-20-24(2)  
**InchiKey:** VWYKFDOLYZLSLQ-UHFFFAOYSA-N  
**Formula:** C26H44O2  
**SMILES:** CCCCCCCCCCCCCCOC(=O)c1ccc(C(C)(C)C)cc1  
**Mol. weight [g/mol]:** 388.63

## Physical Properties

Property code	Value	Unit	Source
gf	39.74	kJ/mol	Joback Method
hf	-608.46	kJ/mol	Joback Method
hfus	52.12	kJ/mol	Joback Method
hvap	84.27	kJ/mol	Joback Method
log10ws	-8.89		Crippen Method
logp	8.232		Crippen Method
mvol	360.880	ml/mol	McGowan Method
pc	900.18	kPa	Joback Method
rinpol	2883.00		NIST Webbook
rinpol	2883.00		NIST Webbook
tb	899.00	K	Joback Method
tc	1102.64	K	Joback Method
tf	496.30	K	Joback Method
vc	1.397	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1192.75	J/molxK	899.00	Joback Method
cpg	1212.55	J/molxK	932.94	Joback Method
cpg	1231.12	J/molxK	966.88	Joback Method
cpg	1248.52	J/molxK	1000.82	Joback Method
cpg	1264.82	J/molxK	1034.76	Joback Method
cpg	1280.11	J/molxK	1068.70	Joback Method
cpg	1294.43	J/molxK	1102.64	Joback Method
dvisc	0.0005181	Paxs	496.30	Joback Method

dvisc	0.0002319	Paxs	563.42	Joback Method
dvisc	0.0001232	Paxs	630.53	Joback Method
dvisc	0.0000739	Paxs	697.65	Joback Method
dvisc	0.0000485	Paxs	764.77	Joback Method
dvisc	0.0000340	Paxs	831.88	Joback Method
dvisc	0.0000252	Paxs	899.00	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406151&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406151&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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