

# Pimelic acid, 2-chlorophenyl tridecyl ester

<b>Inchi:</b>	InChI=1S/C26H41ClO4/c1-2-3-4-5-6-7-8-9-10-11-17-22-30-25(28)20-13-12-14-21-26(29)
<b>InchiKey:</b>	CGHMNEUXOKXSDT-UHFFFAOYSA-N
<b>Formula:</b>	C26H41ClO4
<b>SMILES:</b>	CCCCCCCCCCCCCOC(=O)CCCCC(=O)Oc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	453.05

## Physical Properties

Property code	Value	Unit	Source
gf	-208.95	kJ/mol	Joback Method
hf	-860.25	kJ/mol	Joback Method
hfus	66.52	kJ/mol	Joback Method
hvap	99.10	kJ/mol	Joback Method
log10ws	-8.87		Crippen Method
logp	8.050		Crippen Method
mcvol	380.560	ml/mol	McGowan Method
pc	894.80	kPa	Joback Method
rinpol	2859.00		NIST Webbook
rinpol	2859.00		NIST Webbook
tb	1015.95	K	Joback Method
tc	1245.61	K	Joback Method
tf	595.96	K	Joback Method
vc	1.480	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1269.77	J/molxK	1015.95	Joback Method
cpg	1286.19	J/molxK	1054.23	Joback Method
cpg	1301.01	J/molxK	1092.50	Joback Method
cpg	1314.29	J/molxK	1130.78	Joback Method
cpg	1326.09	J/molxK	1169.06	Joback Method
cpg	1336.48	J/molxK	1207.34	Joback Method
cpg	1345.50	J/molxK	1245.61	Joback Method
dvisc	0.0002398	Paxs	595.96	Joback Method

dvisc	0.0001260	Paxs	665.96	Joback Method
dvisc	0.0000748	Paxs	735.96	Joback Method
dvisc	0.0000486	Paxs	805.95	Joback Method
dvisc	0.0000339	Paxs	875.95	Joback Method
dvisc	0.0000249	Paxs	945.95	Joback Method
dvisc	0.0000191	Paxs	1015.95	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416461&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416461&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/94-990-9/Pimelic-acid-2-chlorophenyl-tridecyl-ester.pdf>

Generated by Cheméo on 2024-04-28 08:37:26.780339898 +0000 UTC m=+16582695.700917226.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.