

# Diethylmalonic acid, 4-bromophenyl tridecyl ester

**Inchi:** InChI=1S/C26H41BrO4/c1-4-7-8-9-10-11-12-13-14-15-16-21-30-24(28)26(5-2,6-3)25(29)  
**InchiKey:** IXHRGLUDTPCJMA-UHFFFAOYSA-N  
**Formula:** C26H41BrO4  
**SMILES:** CCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(Br)cc1  
**Mol. weight [g/mol]:** 497.50

## Physical Properties

Property code	Value	Unit	Source
gf	-179.86	kJ/mol	Joback Method
hf	-826.93	kJ/mol	Joback Method
hfus	60.19	kJ/mol	Joback Method
hvap	99.86	kJ/mol	Joback Method
log10ws	-9.10		Crippen Method
logp	8.015		Crippen Method
mcvol	385.820	ml/mol	McGowan Method
pc	965.67	kPa	Joback Method
rinpol	3089.00		NIST Webbook
rinpol	3089.00		NIST Webbook
tb	1041.45	K	Joback Method
tc	1275.40	K	Joback Method
tf	628.26	K	Joback Method
vc	1.482	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1283.39	J/molxK	1041.45	Joback Method
cpg	1299.48	J/molxK	1080.44	Joback Method
cpg	1314.23	J/molxK	1119.43	Joback Method
cpg	1327.74	J/molxK	1158.42	Joback Method
cpg	1340.11	J/molxK	1197.41	Joback Method
cpg	1351.44	J/molxK	1236.41	Joback Method
cpg	1361.82	J/molxK	1275.40	Joback Method
dvisc	0.0001582	Paxs	628.26	Joback Method

dvisc	0.0000843	Paxs	697.12	Joback Method
dvisc	0.0000503	Paxs	765.99	Joback Method
dvisc	0.0000327	Paxs	834.85	Joback Method
dvisc	0.0000227	Paxs	903.72	Joback Method
dvisc	0.0000166	Paxs	972.58	Joback Method
dvisc	0.0000126	Paxs	1041.45	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369830&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369830&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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