

# Acetic acid, (4-chloro-2-methylphenoxy)-, heptadecyl ester

Inchi:	InChI=1S/C26H43ClO3/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-20-29-26(28)22-30-23
InchiKey:	HGYVHVXYNDBAHX-UHFFFAOYSA-N
Formula:	C26H43ClO3
SMILES:	CCCCCCCCCCCCCCCCOC(=O)Coc1ccc(Cl)cc1C
Mol. weight [g/mol]:	439.07

## Physical Properties

Property code	Value	Unit	Source
gf	-89.66	kJ/mol	Joback Method
hf	-759.14	kJ/mol	Joback Method
hfus	64.53	kJ/mol	Joback Method
hvap	93.02	kJ/mol	Joback Method
log10ws	-9.15		Crippen Method
logp	8.442		Crippen Method
mcvol	378.990	ml/mol	McGowan Method
pc	854.46	kPa	Joback Method
rinpol	1686.00		NIST Webbook
tb	967.06	K	Joback Method
tc	1184.35	K	Joback Method
tf	558.55	K	Joback Method
vc	1.474	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1253.73	J/molxK	967.06	Joback Method
cpg	1271.96	J/molxK	1003.28	Joback Method
cpg	1288.69	J/molxK	1039.49	Joback Method
cpg	1303.97	J/molxK	1075.71	Joback Method
cpg	1317.83	J/molxK	1111.92	Joback Method
cpg	1330.34	J/molxK	1148.14	Joback Method
cpg	1341.52	J/molxK	1184.35	Joback Method
dvisc	0.0002627	Paxs	558.55	Joback Method
dvisc	0.0001369	Paxs	626.63	Joback Method

dvisc	0.0000810	Paxs	694.72	Joback Method
dvisc	0.0000527	Paxs	762.80	Joback Method
dvisc	0.0000368	Paxs	830.89	Joback Method
dvisc	0.0000271	Paxs	898.97	Joback Method
dvisc	0.0000208	Paxs	967.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415170&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415170&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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