# 2-Aminopent-4-enoic acid, N-decyloxycarbonyl-, decyl ester

**Inchi:** InChI=1S/C26H49NO4/c1-4-7-9-11-13-15-17-19-22-30-25(28)24(21-6-3)27-26(29)31-23-

**InchiKey:** ONCSVGPNMNERKX-UHFFFAOYSA-N

**Formula:** C26H49NO4

**SMILES:** C=CCC(N=C(O)OCCCCCCCCCC)C(=O)OCCCCCCCCCC

**Mol. weight [g/mol]:** 439.67

## Physical Properties

| Property code | Value | Unit | Source |
|---|---|---|---|
| hf | -916.64 | kJ/mol | Joback Method |
| hvap | 104.05 | kJ/mol | Joback Method |
| log10ws | -8.10 | | Crippen Method |
| logp | 7.686 | | Crippen Method |
| mcvol | 397.760 | ml/mol | McGowan Method |
| pc | 753.50 | kPa | Joback Method |
| rinpol | 2932.00 | | NIST Webbook |
| rinpol | 2932.00 | | NIST Webbook |
| tb | 1057.97 | K | Joback Method |
| tc | 1321.89 | K | Joback Method |

## Sources

**McGowan Method:** http://link.springer.com/article/10.1007/BF02311772

**NIST Webbook:** http://webbook.nist.gov/cgi/cbook.cgi?ID=U393176&Units=SI

**Crippen Method:** http://pubs.acs.org/doi/abs/10.1021/ci990307l

**Crippen Method:** https://www.chemeo.com/doc/models/crippen_log10ws

**Joback Method:** https://en.wikipedia.org/wiki/Joback_method

## Legend

**hf:** Enthalpy of formation at standard conditions

**hvap:** Enthalpy of vaporization at standard conditions

| | |
|---|---|
| **log10ws:** | Log10 of Water solubility in mol/l |
| **logp:** | Octanol/Water partition coefficient |
| **mcvol:** | McGowan's characteristic volume |
| **pc:** | Critical Pressure |
| **rinpol:** | Non-polar retention indices |
| **tb:** | Normal Boiling Point Temperature |
| **tc:** | Critical Temperature |