

# Pentane, 1,5-diiodo-

<b>Other names:</b>	1,5-Diiodopentane Pentamethylene diiodide
<b>Inchi:</b>	InChI=1S/C5H10I2/c6-4-2-1-3-5-7/h1-5H2
<b>InchiKey:</b>	IAEOYUUPFYJXHN-UHFFFAOYSA-N
<b>Formula:</b>	C5H10I2
<b>SMILES:</b>	ICCCCCI
<b>Mol. weight [g/mol]:</b>	323.94
<b>CAS:</b>	628-77-3

## Physical Properties

Property code	Value	Unit	Source
gf	107.46	kJ/mol	Joback Method
hf	7.21	kJ/mol	Joback Method
hfus	17.52	kJ/mol	Joback Method
hvap	45.47	kJ/mol	Joback Method
log10ws	-3.81		Crippen Method
logp	3.027		Crippen Method
mcvol	132.950	ml/mol	McGowan Method
pc	3295.37	kPa	Joback Method
tb	500.08	K	Joback Method
tc	739.07	K	Joback Method
tf	262.23	K	Joback Method
vc	0.491	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	213.41	J/mol×K	500.08	Joback Method
cpg	222.70	J/mol×K	539.91	Joback Method
cpg	231.35	J/mol×K	579.74	Joback Method
cpg	239.40	J/mol×K	619.58	Joback Method
cpg	246.90	J/mol×K	659.41	Joback Method
cpg	253.90	J/mol×K	699.24	Joback Method
cpg	260.46	J/mol×K	739.07	Joback Method

dvisc	0.0056592	Paxs	262.23	Joback Method
dvisc	0.0026807	Paxs	301.87	Joback Method
dvisc	0.0015104	Paxs	341.51	Joback Method
dvisc	0.0009588	Paxs	381.15	Joback Method
dvisc	0.0006631	Paxs	420.80	Joback Method
dvisc	0.0004886	Paxs	460.44	Joback Method
dvisc	0.0003779	Paxs	500.08	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	422.20	K	2.70	NIST Webbook
tbrp	374.70	K	0.40	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.33708e+01
Coeff. B	-4.17109e+03
Coeff. C	-8.75860e+01
Temperature range (K), min.	406.40
Temperature range (K), max.	605.14

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C628773&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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