

# Sebacic acid, 2-iodobenzyl octyl ester

<b>Inchi:</b>	InChI=1S/C25H39IO4/c1-2-3-4-5-10-15-20-29-24(27)18-11-8-6-7-9-12-19-25(28)30-21-2
<b>InchiKey:</b>	LCGGMJZULKFCCG-UHFFFAOYSA-N
<b>Formula:</b>	C25H39IO4
<b>SMILES:</b>	CCCCCCCCOC(=O)CCCCCCCC(=O)OCc1ccccc1I
<b>Mol. weight [g/mol]:</b>	530.48

## Physical Properties

Property code	Value	Unit	Source
gf	-147.32	kJ/mol	Joback Method
hf	-747.00	kJ/mol	Joback Method
hfus	64.14	kJ/mol	Joback Method
hvap	101.87	kJ/mol	Joback Method
log10ws	-8.76		Crippen Method
logp	7.359		Crippen Method
mcvol	380.050	ml/mol	McGowan Method
pc	957.32	kPa	Joback Method
rinpol	3250.00		NIST Webbook
rinpol	3250.00		NIST Webbook
tb	1048.78	K	Joback Method
tc	1284.22	K	Joback Method
tf	612.83	K	Joback Method
vc	1.464	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1232.45	J/molxK	1048.78	Joback Method
cpg	1293.32	J/molxK	1244.98	Joback Method
cpg	1283.77	J/molxK	1205.74	Joback Method
cpg	1272.98	J/molxK	1166.50	Joback Method
cpg	1260.88	J/molxK	1127.26	Joback Method
cpg	1247.39	J/molxK	1088.02	Joback Method
cpg	1301.71	J/molxK	1284.22	Joback Method
dvisc	0.0000176	Paxs	1048.78	Joback Method

dvisc	0.0000230	Paxs	976.12	Joback Method
dvisc	0.0000313	Paxs	903.46	Joback Method
dvisc	0.0000449	Paxs	830.80	Joback Method
dvisc	0.0000691	Paxs	758.15	Joback Method
dvisc	0.0001165	Paxs	685.49	Joback Method
dvisc	0.0002225	Paxs	612.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U380675&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U380675&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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